QP CODE:111006 (2012 - Scheme)	Reg.No:
--------------------------------	---------

First Year B.Pharm Degree Examinations - September 2013

(2012 - Scheme)

PHARMACEUTICAL CHEMISTRY - I

(Inorganic & Physical Chemistry)

Time: 3 Hours Total Marks: 100

- Answer all Questions.
- Write equations wherever necessary.

Essay (3x10=30)

- 1. What is ORS. List the contents and role of each ion. Where it is useful
- 2. Define adsorbents and protectives giving examples. Explain the advantages of antacid combination therapy with examples.
- 3. What are astringents. Mention examples and its uses. How do you prepare and assay hydrogen peroxide

Short notes (14x5=70)

- 4. Explain the uses of magnesium tri silicate and aluminium hydroxide gel with its chemical formula
- 5. Importance of heavy metal limit test. Explain the procedure to carry out the same.
- 6. Role of fluorides ions in dental caries. How do you prepare sodium fluoride
- 7. How do you prepare and assay inhalant oxygen
- 8. Iron compounds used as pharmaceuticals
- 9. Discuss the physiological role of copper and iodine
- 10. Sodium lauryl sulphate and its importance in pharmacy
- 11. Discuss different qualities of water in IP
- 12. What are inorganic expectorant. List examples and its uses
- 13. How does 1,10-phenanthroline help in pharmaceutical analysis
- 14. Explain the colligative properties with examples
- 15. What is kinetic theory of gas and its significance
- 16. Sources of impurities in pharmaceuticals
- 17. What are the units for measuring radioactivity. List the uses of iodine-131
